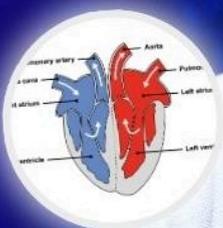
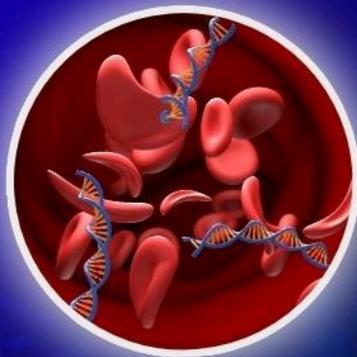


REPEATED QUESTIONS

CHEMISTRY



FOR CUET 2026

Chemistry

1.

Geometry of $[\text{Ni}(\text{CN})_4]^{2-}$ and $[\text{NiCl}_4]^{2-}$ are

1. Tetrahedral, low spin and square planar, high spin respectively
2. Tetrahedral, high spin and square planar, low spin respectively
3. Square planar, low spin and tetrahedral, high spin respectively
4. Square planar, high spin and tetrahedral, low spin respectively

[Shift 27/05/2023 3:30 PM - 6:30 PM, Qno: 5]

Similar Questions:

Geometry of $[\text{Ni}(\text{CN})_4]^{2-}$ and $[\text{NiCl}_4]^{2-}$ are

1. Tetrahedral, low spin and square planar, high spin respectively
2. Tetrahedral, high spin and square planar, low spin respectively
3. Square planar, low spin and tetrahedral, high spin respectively
4. Square planar, high spin and tetrahedral, low spin respectively

[shift-02-06-2025-9:00AM-12:00PM, Qno: 1]

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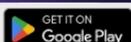
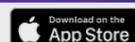
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2.

The standard reduction potentials are given as: $E^\circ_{K^+/K} = -2.93$, $E^\circ_{Ag^+/Ag} = +0.80$, $E^\circ_{Cu^{2+}/Cu} = +0.34$, $E^\circ_{Ni^{2+}/Ni} = -0.25$, $E^\circ_{Li^+/Li} = -3.05$, V

Arrange the following metals in increasing order of their reducing power: A.

B. Ag C. Cu D. Ni E. Li Choose the correct answer from the options given below: 1.

B < C < A < D < E

2. B < C < E < A < D

3. B < C < D < A < E

4. B < C < A < E < D

[Shift 23/05/2023 3:30 PM - 6:30 PM, Qno: 7]

Similar Questions:

The standard reduction potentials are given as: $E^\circ_{K^+/K} = -2.93$, $E^\circ_{Ag^+/Ag} =$

$+0.80$, $E^\circ_{Cu^{2+}/Cu} = +0.34$, $E^\circ_{Ni^{2+}/Ni} = -0.25$, $E^\circ_{Li^+/Li} = -3.05$, V

Arrange the following metals in increasing order of their reducing power: A. KB. Ag C. Cu D. Ni E. Li Choose

the correct answer from the options given below: 1. B < C < A < D < E

2. B < C < E < A < D

3. B < C < D < A < E

4. B < C < A < E < D

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 4]

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3.

Which one of the following ligands forms a chelate?

1. Cyanide
2. Water
3. Oxalate
4. Chloride

[Shift 23/05/2023 3:30 PM - 6:30 PM, Qno: 14]

Similar Questions:

Which one of the following ligands forms a chelate?

1. Cyanide
2. Water
3. Oxalate
4. Chloride

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 7]

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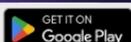
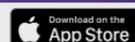
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4.

In Cu - Ag electrochemical cell Cu acts as

1. Anode
2. Cathode
3. Salt bridge
4. Electrolyte

[Shift 23/05/2023 3:30 PM - 6:30 PM, Qno: 17]

Similar Questions:

In Cu - Ag electrochemical cell Cu acts as

1. Anode
2. Cathode
3. Salt bridge
4. Electrolyte

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 10]

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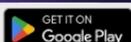
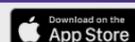
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5.

The complex $[\text{Co}(\text{NH}_3)_4(\text{NO}_2)_2]\text{Cl}$ exhibits:

- A. Coordination isomerism
- B. Linkage isomerism
- C. Solvate isomerism
- D. Ionisation isomerism
- E. Geometrical isomerism

Choose the **correct** answer from the options given below:

- 1. B, D and E only
- 2. B, A and D only
- 3. C, A and D only
- 4. D, E and C only

[Shift 23/05/2023 3:30 PM - 6:30 PM, Qno: 23]

Similar Questions:

The complex $[\text{Co}(\text{NH}_3)_4(\text{NO}_2)_2]\text{Cl}$ exhibits:

- A. Coordination isomerism
- B. Linkage isomerism
- C. Solvate isomerism
- D. Ionisation isomerism
- E. Geometrical isomerism

Choose the correct answer from the options given below:

- 1. B, D and E only
- 2. B, A and D only
- 3. C, A and D only
- 4. D, E and C only

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 13]

The complex $[\text{Co}(\text{NH}_3)_4(\text{NO}_2)_2]\text{Cl}$ exhibits:

- A. Coordination isomerism
- B. Linkage isomerism

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- C. Solvate isomerism
- D. Ionisation isomerism
- E. Geometrical isomerism

Choose the correct answer from the options given below:

- 1. B, D and E only
- 2. B, A and D only
- 3. C, A and D only
- 4. D, E and C only

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 179]

6.

If we plot graph for first order reaction involving reactant R, then it will be a straight line if plotted for :

- 1. $[R] / [R]$ against time, t
- 2. $[R]$. against time, t
- 3. $\log [R] / [R]$. against time, t
- 4. $[R]$ against time, t

[Shift 23/05/2023 3:30 PM - 6:30 PM, Qno: 35]

Similar Questions:

If we plot graph for first order reaction involving reactant R, then it will be a straight line if plotted for :

- 1. $[R] / [R]$ against time, t
- 2. $[R]$. against time, t
- 3. $\log [R] / [R]$. against time, t
- 4. $[R]$ against time, t

[Shift 12/06/2023 12:00 PM - 2:00 PM, Qno: 20]

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7.

The correct order for increasing the boiling point of hydrides (given below) of group 15 is:

- (A) NH_3
- (B) PH_3
- (C) AsH_3
- (D) SbH_3
- (E) BiH_3

Choose the correct answer from the options given below :

- (1) $A < B < C < D < E$
- (2) $A < E < D < C < B$
- (3) $B < C < A < D < E$
- (4) $B < C < D < A < E$

[Shift 24/05/2023 3:30 PM - 6:30 PM, Qno: 34]

Similar Questions:

The correct order of increasing boiling points of hydrides (given below) of group 15 is:

- (A) NH_3
- (B) PH_3
- (C) AsH_3
- (D) SbH_3
- (E) BiH_3

Choose the correct answer from the options given below :

- (1) $A < B < C < D < E$
- (2) $A < E < D < C < B$
- (3) $B < C < A < D < E$
- (4) $B < C < D < A < E$

[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 24]

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8.

To prepare 250 mL standard solution of M/20 NaCl, amount of NaCl required is :

- (1) 7.3 g
- (2) 0.73 g
- (3) 0.073 g
- (4) 73 g

[Shift 24/05/2023 3:30 PM - 6:30 PM, Qno: 40]

Similar Questions:

To prepare 250 mL standard solution of M/20 NaCl, amount of NaCl required is :

- (1) 7.3 g
- (2) 0.73 g
- (3) 0.073 g
- (4) 73 g

[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 36]

To prepare 250 mL standard solution of M/20 NaCl, amount of NaCl required is :

- (1) 7.3 g
- (2) 0.73 g
- (3) 0.073 g
- (4) 73 g

[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 103]

To prepare 250 mL standard solution of M/20 NaCl, amount of NaCl required is :

- (1) 7.3 g
- (2) 0.73 g
- (3) 0.073 g
- (4) 73 g

[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 179]

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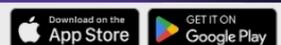
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9.

In which case there is a change in oxidation number? (1) Aqueous solution of CrO_4^{2-} is acidified. (2) SO_2 gas is passed through acidified $\text{Cr}_2\text{O}_7^{2-}$ solution. (3) $\text{Cr}_2\text{O}_7^{2-}$ solution is made alkaline. (4) CrO_2Cl_2 is dissolved in NaOH .

[Shift 30/05/2023 12:00 PM - 2:00 PM, Qno: 31]

Similar Questions:

In which of the following cases is there a change in oxidation number?(1) Aqueous solution of CrO_4^{2-} is acidified(2) SO_2 gas is passed through acidified $\text{Cr}_2\text{O}_7^{2-}$ solution(3) $\text{Cr}_2\text{O}_7^{2-}$ solution is made alkaline(4) CrO_2Cl_2 is dissolved in NaOH

[Shift 07/06/2023 3:30 PM - 6:30 PM, Qno: 17]

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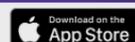
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10.

Which of the following statements is not correct?

(1) La is an element of transition series rather than Lanthanoid series (2) $\text{La}(\text{OH})_3$ is less basic than $\text{Lu}(\text{OH})_3$ (3) In Lanthanoid series, ionic radius of Ln^{3+} ion decreases (4) Atomic radii of Zr and Hf are same because of Lanthanoid contraction

[Shift 22/05/2023 3:30 PM - 6:30 PM, Qno: 37]

Similar Questions:

Which of the following is not correct ?

A. $\text{La}(\text{OH})_3$ is less basic than $\text{Lu}(\text{OH})_3$ B. In Lanthanoid series, ionic radius of M^{3+} ion decreases from La to Lu. C. La is an element of transition series rather than a Lanthanoid series. D. Atomic radii of Zr and Hf are the same because of Lanthanoid contraction.

Choose the correct answer from the options given below:

1. B and C only
2. C and A only
3. A only
4. D and B only

[Shift 08/06/2023 3:30 PM - 6:30 PM, Qno: 10]

Which of the following is not correct ?

A. $\text{La}(\text{OH})_3$ is less basic than $\text{Lu}(\text{OH})_3$ B. In Lanthanoid series, ionic radius of M^{3+} ion decreases from La to Lu. C. La is an element of transition series rather than a Lanthanoid series. D. Atomic radii of Zr and Hf are the same because of Lanthanoid contraction.

Choose the correct answer from the options given below:

1. B and C only
2. C and A only
3. A only
4. D and B only

[Shift 08/06/2023 3:30 PM - 6:30 PM, Qno: 125]

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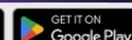
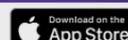
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11.

The rate of a gaseous reaction becomes half when the volume of the vessel is doubled. The order of the reaction is

- (A) Zero
- (B) One
- (C) Two
- (D) Three

[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 5]

Similar Questions:

The rate of a gaseous reaction becomes half when the volume of the vessel is doubled. The order of the reaction is

- (A) Zero
- (B) One
- (C) Two
- (D) Three

[Chemistry 30th Aug Slot 2 , Qno: 13]

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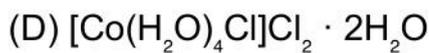
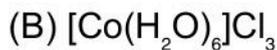
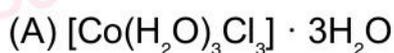


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12.

Which one of the following 1 molal aqueous solutions will show maximum freezing point depression?

Options:

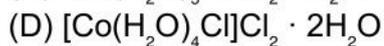
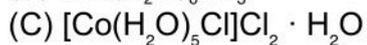
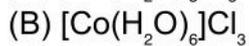
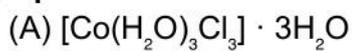


[Shift 28/05/2023 8:30 AM - 10:30 AM, Qno: 34]

Similar Questions:

Which one of the following 1 molal aqueous solutions will show maximum freezing point depression?

Options:



[Chemistry 30th Aug Slot 2 , Qno: 5]

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13.

Arrange the following electrolytes in the increasing order of van't Hoff factor (i) for complete dissociation of solute.

(A) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ (B) MgCl_2 (C) $\text{K}_4[\text{Fe}(\text{CN})_6]$ (D) KCl

Choose the correct answer from the options given below:

1. (A), (B), (C), (D)
2. (D), (B), (A), (C)
3. (D), (A), (B), (C)
4. (C), (A), (B), (D)

[Shift 27/05/2023 8:30 AM - 11:30 AM, Qno: 28]

Similar Questions:

Arrange the following electrolytes in the increasing order of van't Hoff factor (i) for complete dissociation of solute.

(A) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ (B) MgCl_2 (C) $\text{K}_4[\text{Fe}(\text{CN})_6]$ (D) KCl

Choose the correct answer from the options given below:

1. (A), (B), (C), (D)
2. (D), (B), (A), (C)
3. (D), (A), (B), (C)
4. (C), (A), (B), (D)

[Shift 09/06/2023 3:30 PM - 6:30 PM, Qno: 37]

Arrange the following electrolytes in the increasing order of van't Hoff factor (i) for complete dissociation of solute.

(A) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ (B) MgCl_2 (C) $\text{K}_4[\text{Fe}(\text{CN})_6]$ (D) KCl

Choose the correct answer from the options given below:

1. (A), (B), (C), (D)
2. (D), (B), (A), (C)
3. (D), (A), (B), (C)
4. (C), (A), (B), (D)

[Shift 10/06/2023 3:30 PM - 6:30 PM, Qno: 8]

Arrange the following electrolytes in the increasing order of van't Hoff factor (i) for complete dissociation of solute.

(A) $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ (B) MgCl_2 (C) $\text{K}_4[\text{Fe}(\text{CN})_6]$ (D) KCl

Choose the correct answer from the options given below:

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1. (A), (B), (C), (D)
2. (D), (B), (A), (C)
3. (D), (A), (B), (C)
4. (C), (A), (B), (D)

[shift-02-06-2025-3:00PM-6:00PM, Qno: 1]

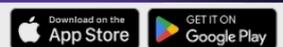
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14.

The amines which are soluble in water

(A) Methanamine(B) Ethanamine(C) Propanamine(D) Benzylamine

Choose the **correct** answer from the options given below:

1. (A), (B) and (D) only
2. (B), (C) and (D) only
3. (A), (B), (C) and (D)
4. (A), (B) and (C) only

[Shift 27/05/2023 8:30 AM - 11:30 AM, Qno: 29]

Similar Questions:

The amines which are soluble in water

(A) Methanamine(B) Ethanamine(C) Propanamine(D) Benzylamine

Choose the **correct** answer from the options given below:

1. (A), (B) and (D) only
2. (B), (C) and (D) only
3. (A), (B), (C) and (D)
4. (A), (B) and (C) only

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 33]

The amines which are soluble in water

(A) Methanamine(B) Ethanamine(C) Propanamine(D) Benzylamine

Choose the correct answer from the options given below:

1. (A), (B) and (D) only
2. (B), (C) and (D) only
3. (A), (B), (C) and (D)
4. (A), (B) and (C) only

[Shift 09/06/2023 3:30 PM - 6:30 PM, Qno: 28]

The amines which are soluble in water

(A) Methanamine(B) Ethanamine(C) Propanamine(D) Benzylamine

Choose the correct answer from the options given below:

1. (A), (B) and (D) only
2. (B), (C) and (D) only
3. (A), (B), (C) and (D)
4. (A), (B) and (C) only

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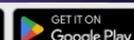
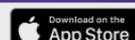
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[Shift 10/06/2023 3:30 PM - 6:30 PM, Qno: 15]

The amines which are soluble in water

(A) Methanamine(B) Ethanamine(C) Propanamine(D) Benzylamine

Choose the **correct** answer from the options given below:

1. (A), (B) and (D) only
2. (B), (C) and (D) only
3. (A), (B), (C) and (D)
4. (A), (B) and (C) only

[shift-02-06-2025-3:00PM-6:00PM, Qno: 9]

15.

Which of the following vitamin is water soluble ?

1. Vitamin A
2. Vitamin D
3. Vitamin B
4. Vitamin E

[Shift 27/05/2023 8:30 AM - 11:30 AM, Qno: 34]

Similar Questions:

Which of the following vitamins is soluble in water?

1. Vitamin E
2. Vitamin D
3. Vitamin B
4. Vitamin K

[shift-29-05-2025-3:00PM-6:00PM, Qno: 19]

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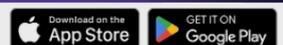
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16.

- Magnetic moment values for $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ are1. 1.74 BM and 5.92 respectively
2. 5.92 BM and 1.74 respectively
3. 1.74 BM for both
4. 5.92 BM for both

[Shift 27/05/2023 8:30 AM - 11:30 AM, Qno: 36]

Similar Questions:

- Magnetic moment values for $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ are1. 1.74 BM and 5.92 respectively
2. 5.92 BM and 1.74 respectively
3. 1.74 BM for both
4. 5.92 BM for both

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 39]

- Magnetic moment values for $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ are1. 1.74 BM and 5.92 respectively
2. 5.92 BM and 1.74 respectively
3. 1.74 BM for both
4. 5.92 BM for both

[Shift 09/06/2023 3:30 PM - 6:30 PM, Qno: 34]

- Magnetic moment values for $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ are1. 1.74 BM and 5.92 respectively
2. 5.92 BM and 1.74 respectively
3. 1.74 BM for both
4. 5.92 BM for both

[Shift 09/06/2023 3:30 PM - 6:30 PM, Qno: 103]

- Magnetic moment values for $[\text{Fe}(\text{H}_2\text{O})_6]^{3+}$ and $[\text{Fe}(\text{CN})_6]^{3-}$ are1. 1.74 BM and 5.92 respectively
2. 5.92 BM and 1.74 respectively
3. 1.74 BM for both
4. 5.92 BM for both

[shift-02-06-2025-3:00PM-6:00PM, Qno: 17]

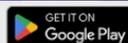
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17.

The mass of NaOH required to make a 500 mL solution of 0.5 molarity is

1. 15 g
2. 20 g
3. 40 g
4. 10 g

[Shift 27/05/2023 8:30 AM - 11:30 AM, Qno: 37]

Similar Questions:

The mass of NaOH required to make a 500 mL solution of 0.5 molarity is

1. 15 g
2. 20 g
3. 40 g
4. 10 g

[shift-02-06-2025-3:00PM-6:00PM, Qno: 21]

18.

Type of RNA which does not exist is

1. m-RNA
2. s-RNA
3. r-RNA
4. t-RNA

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 8]

Similar Questions:

Type of RNA which does not exist is

1. m-RNA
2. s-RNA
3. r-RNA
4. t-RNA

[shift-02-06-2025-3:00PM-6:00PM, Qno: 79]

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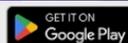
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19.

Hybridization of carbon atoms attached with halogens in haloalkanes & haloarenes are

1. sp^2 & sp^2 hybridized respectively
2. sp^3 & sp^2 hybridized respectively
3. sp^3 & sp^3 hybridized respectively
4. sp^2 & sp^3 hybridized respectively

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 10]

Similar Questions:

Hybridization of carbon atoms attached with halogens in haloalkanes & haloarenes are

1. sp^2 & sp^2 hybridized respectively
2. sp^3 & sp^2 hybridized respectively
3. sp^3 & sp^3 hybridized respectively
4. sp^2 & sp^3 hybridized respectively

[shift-02-06-2025-3:00PM-6:00PM, Qno: 77]

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20.

What will happen during the electrolysis of aqueous solution of CuSO_4 using platinum electrodes? A. Copper will deposit at cathode. B. Copper will deposit at anode. C. Oxygen will be released at anode. D. Oxygen will be released at cathode. Choose the correct answer from the options given below:

1. B and D
2. C and D
3. A and C
4. B and C

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 13]

Similar Questions:

What will happen during the electrolysis of aqueous solution of CuSO_4 by using platinum electrodes?

A. Copper will deposit at cathode. B. Copper will deposit at anode. C. Oxygen will be released at anode. D. Copper will dissolve at anode.

Choose the correct answer from the options given below:

1. C and D only
2. A and C only
3. A and B only
4. B and D only

[Shift 08/06/2023 3:30 PM - 6:30 PM, Qno: 24]

What will happen during the electrolysis of aqueous solution of CuSO_4 by using platinum electrodes?

A. Copper will deposit at cathode. B. Copper will deposit at anode. C. Oxygen will be released at anode. D. Copper will dissolve at anode.

Choose the correct answer from the options given below:

1. C and D only
2. A and C only
3. A and B only
4. B and D only

[Shift 08/06/2023 3:30 PM - 6:30 PM, Qno: 109]

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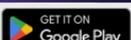
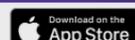
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21.

Which statement is correct for solutions?

1. Ideal solutions do not obey Raoult's law.
2. For an ideal solution, $H_{mix} > 0$ and $V_{mix} > 0$.
3. If $H_{mix} > 0$ and $V_{mix} > 0$, then solutions show negative deviations.
4. If $H_{mix} > 0$ and $V_{mix} > 0$, then solutions show positive deviations.

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 18]

Similar Questions:

Which statement is correct for solutions?

1. Ideal solutions do not obey Raoult's law.
2. For an ideal solution, $H_{mix} > 0$ and $V_{mix} > 0$.
3. If $H_{mix} > 0$ and $V_{mix} > 0$, then solutions show negative deviations.
4. If $H_{mix} > 0$ and $V_{mix} > 0$, then solutions show positive deviations.

[shift-02-06-2025-3:00PM-6:00PM, Qno: 73]

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22.

Arrange the following in order of **increasing boiling points**

- (A) 1,4-Dichlorobenzene
- (B) 1,3-Dichlorobenzene
- (C) 1,2-Dichlorobenzene
- (D) Chlorobenzene

Choose the **correct** answer from the options given below:

- 1. (D), (C), (B), (A)
- 2. (A), (B), (C), (D)
- 3. (B), (A), (D), (C)
- 4. (D), (B), (A), (C)

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 19]

Similar Questions:

Arrange the following in order of **increasing boiling points**

- (A) 1,4-Dichlorobenzene
- (B) 1,3-Dichlorobenzene
- (C) 1,2-Dichlorobenzene
- (D) Chlorobenzene

Choose the **correct** answer from the options given below:

- 1. (D), (C), (B), (A)
- 2. (A), (B), (C), (D)
- 3. (B), (A), (D), (C)
- 4. (D), (B), (A), (C)

[shift-02-06-2025-3:00PM-6:00PM, Qno: 75]

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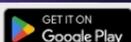
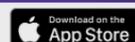
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23.

The product obtained, when starch is hydrolyzed with dilute sulphuric acid at 393 K under pressure, is $(C_6H_{10}O_5)_n + nH_2O$ [393 K; 2-3 atm]H⁺ ? Options:

1. Glucose only
2. A mixture of glucose and fructose
3. Fructose only
4. Gluconic acid

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 21]

Similar Questions:

The product obtained, when starch is hydrolyzed with dilute sulphuric acid at 393 K under pressure, is $(C_6H_{10}O_5)_n + nH_2O$ [393 K; 2-3 atm]H⁺ ? Options:

1. Glucose only
2. A mixture of glucose and fructose
3. Fructose only
4. Gluconic acid

[shift-02-06-2025-3:00PM-6:00PM, Qno: 71]

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24.

Arrange the following elements in decreasing order of their radii: (A) La (B) Pr (C) Gd (D) Yb

Choose the correct answer from the options given below:

1. (B), (A), (C), (D)
2. (A), (B), (C), (D)
3. (D), (A), (C), (B)
4. (A), (B), (D), (C)

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 22]

Similar Questions:

Arrange the following elements in decreasing order of their radii: (A) La (B) Pr (C) Gd (D) Yb

Choose the correct answer from the options given below:

1. (B), (A), (C), (D)
2. (A), (B), (C), (D)
3. (D), (A), (C), (B)
4. (A), (B), (D), (C)

[shift-02-06-2025-3:00PM-6:00PM, Qno: 67]

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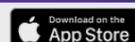
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25.

Identify Identify the optically active compounds from the following: (A) $[\text{Co}(\text{en})_3]^{3+}$ (B) $\text{trans-}[\text{Co}(\text{en})_2\text{Cl}_2]^+$ (C) $\text{cis-}[\text{Co}(\text{en})_2\text{Cl}_2]^+$ (D) $[\text{Cr}(\text{NH}_3)_5\text{Cl}]$ Choose the correct answer from the options given below:

1. (A), (C) and (D) only
2. (A), (B) and (D) only
3. (A), and (C) only
4. (B) and (D) only

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 24]

Similar Questions:

Identify Identify the optically active compounds from the following: (A) $[\text{Co}(\text{en})_3]^{3+}$ (B) $\text{trans-}[\text{Co}(\text{en})_2\text{Cl}_2]^+$ (C) $\text{cis-}[\text{Co}(\text{en})_2\text{Cl}_2]^+$ (D) $[\text{Cr}(\text{NH}_3)_5\text{Cl}]$ Choose the correct answer from the options given below:

1. (A), (C) and (D) only
2. (A), (B) and (D) only
3. (A), and (C) only
4. (B) and (D) only

[shift-02-06-2025-3:00PM-6:00PM, Qno: 65]

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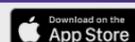
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26.

Which of the following is correct about the nature of ligand?

1. CN^- is an ambidentate ligand
2. Ethylenediaminetetraacetate ion (EDTA^{4-}) is tetradentate ligand
3. CH_3NH_2 is a bidentate ligand
4. $\text{C}_2\text{O}_4^{2-}$ is not a Chelating ligand

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 29]

Similar Questions:

Which of the following is correct about the nature of ligand?

1. CN^- is an ambidentate ligand
2. Ethylenediaminetetraacetate ion (EDTA^{4-}) is tetradentate ligand
3. CH_3NH_2 is a bidentate ligand
4. $\text{C}_2\text{O}_4^{2-}$ is not a Chelating ligand

[shift-02-06-2025-3:00PM-6:00PM, Qno: 61]

27.

Arrange the following in increasing order of their pK_a values. (A) o-nitrophenol (B) m-nitrophenol (C) Phenol (D) Ethanol Choose the correct answer from the options given below:

1. (A), (B), (C), (D)
2. (B), (A), (C), (D)
3. (D), (C), (B), (A)
4. (B), (A), (C), (D)

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 35]

Similar Questions:

Arrange the following in increasing order of their pK_a values. (A) o-nitrophenol (B) m-nitrophenol (C) Phenol (D) Ethanol Choose the correct answer from the options given below:

1. (A), (B), (C), (D)
2. (B), (A), (C), (D)
3. (D), (C), (B), (A)
4. (B), (A), (C), (D)

[shift-02-06-2025-3:00PM-6:00PM, Qno: 11]

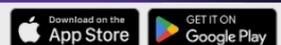
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28.

Which of the following compounds will not undergo azo coupling reaction with benzene diazonium chloride?

1. Aniline
2. Nitrobenzene
3. Phenol
4. Anisole

[Shift 31/05/2023 3:30 PM - 6:30 PM, Qno: 40]

Similar Questions:

Which of the following compounds will not undergo azo coupling with benzene diazonium chloride?

1. Anisole
2. Phenol
3. Aniline
4. Nitrobenzene

[shift-23-05-2025-3:00PM-6:00PM, Qno: 89]

Which of the following compounds will not give azo coupling reaction with benzene diazonium chloride?

1. Nitrobenzene
2. Aniline
3. o-Toluidine
4. Phenol

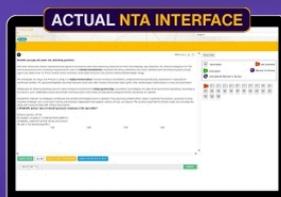
[shift-13-05-2025-9:00AM-12:00PM, Qno: 31]

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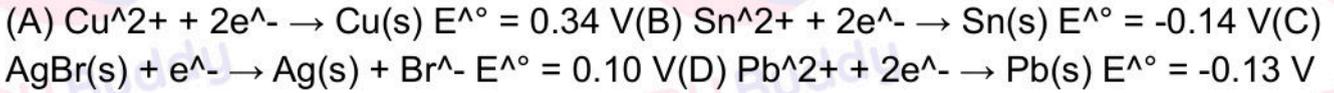
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29.

Arrange the following electrodes in decreasing order of the strength of the respective ions as reducing agent.



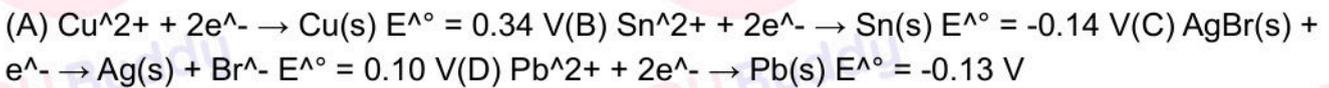
Choose the correct answer from the options given below:

1. (B), (D), (C), (A)
2. (A), (C), (D), (B)
3. (C), (A), (B), (D)
4. (D), (B), (A), (C)

[Shift 01/06/2023 3:30 PM - 6:30 PM, Qno: 1]

Similar Questions:

Arrange the following electrodes in decreasing order of the strength of the respective ions as reducing agent.

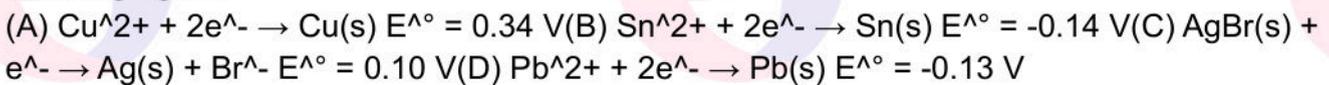


Choose the correct answer from the options given below:

1. (B), (D), (C), (A)
2. (A), (C), (D), (B)
3. (C), (A), (B), (D)
4. (D), (B), (A), (C)

[Shift 09/06/2023 3:30 PM - 6:30 PM, Qno: 35]

Arrange the following electrodes in decreasing order of the strength of the respective ions as reducing agent.



Choose the correct answer from the options given below:

1. (B), (D), (C), (A)
2. (A), (C), (D), (B)
3. (C), (A), (B), (D)
4. (D), (B), (A), (C)

[shift-02-06-2025-3:00PM-6:00PM, Qno: 19]

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30.

Identify the carbohydrate which is not a polysaccharide

1. Starch
2. Cellulose
3. Canesugar
4. Glycogen

[Shift 01/06/2023 3:30 PM - 6:30 PM, Qno: 2]

Similar Questions:

Identify the carbohydrate which is not a polysaccharide

1. Starch
2. Cellulose
3. Canesugar
4. Glycogen

[shift-02-06-2025-3:00PM-6:00PM, Qno: 23]

31.

Osmotic pressure of 0.02 M urea solution at 27°C, is (given $R = 0.0821 \text{ atm L mol}^{-1} \text{ K}^{-1}$)

1. 0.4926 atm
2. 0.8321 atm
3. 1.50 atm
4. 2.00 atm

[Shift 01/06/2023 3:30 PM - 6:30 PM, Qno: 4]

Similar Questions:

Osmotic pressure of 0.02 M urea solution at 27°C, is (given $R = 0.0821 \text{ atm L mol}^{-1} \text{ K}^{-1}$)

1. 0.4926 atm
2. 0.8321 atm
3. 1.50 atm
4. 2.00 atm

[shift-02-06-2025-3:00PM-6:00PM, Qno: 27]

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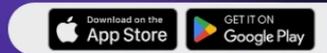
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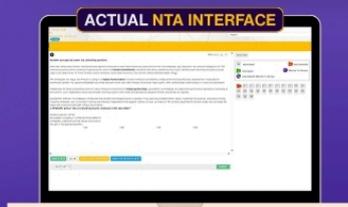


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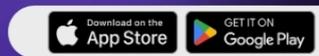
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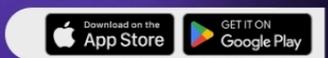
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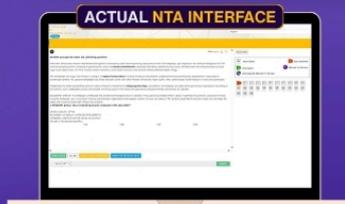
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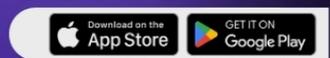
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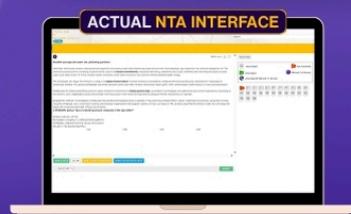
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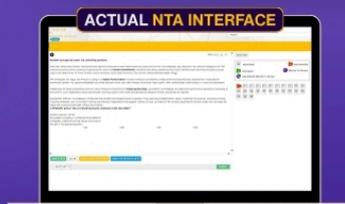
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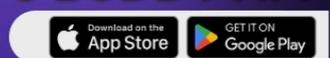
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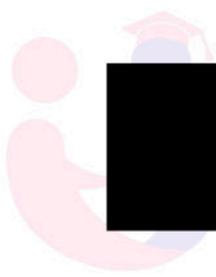
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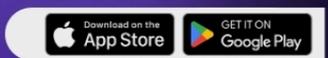
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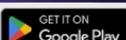
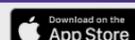
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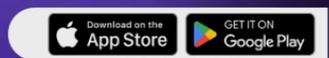
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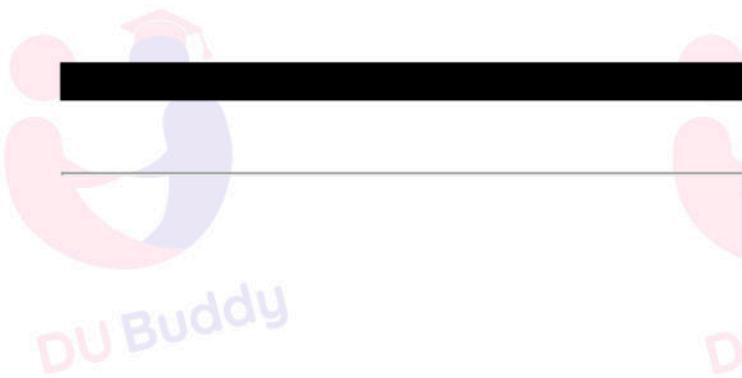
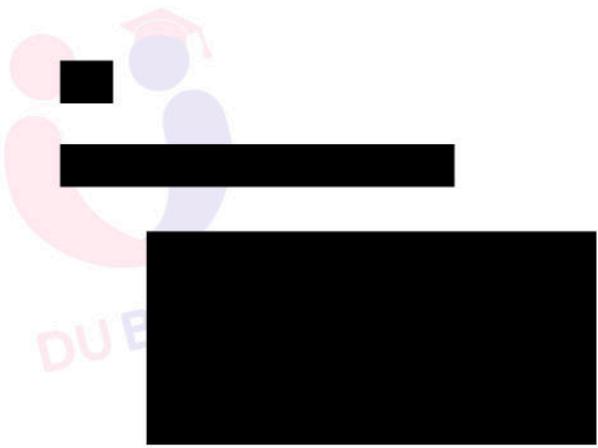
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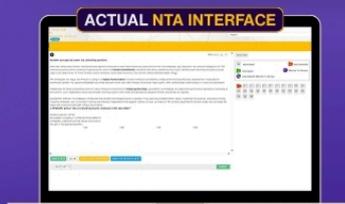
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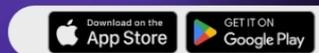
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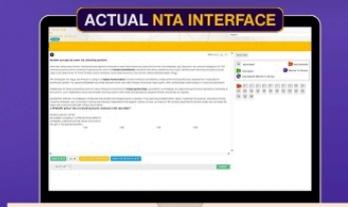
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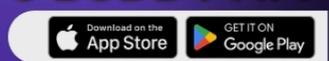
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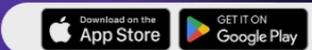
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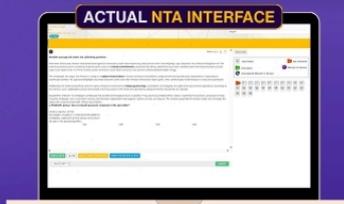
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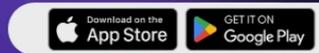
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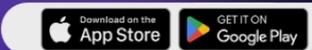
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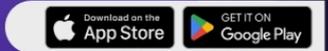
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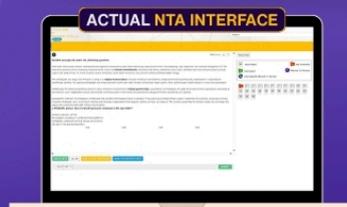
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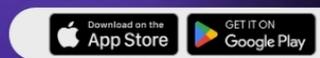
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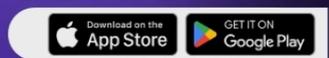
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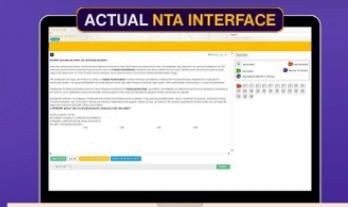
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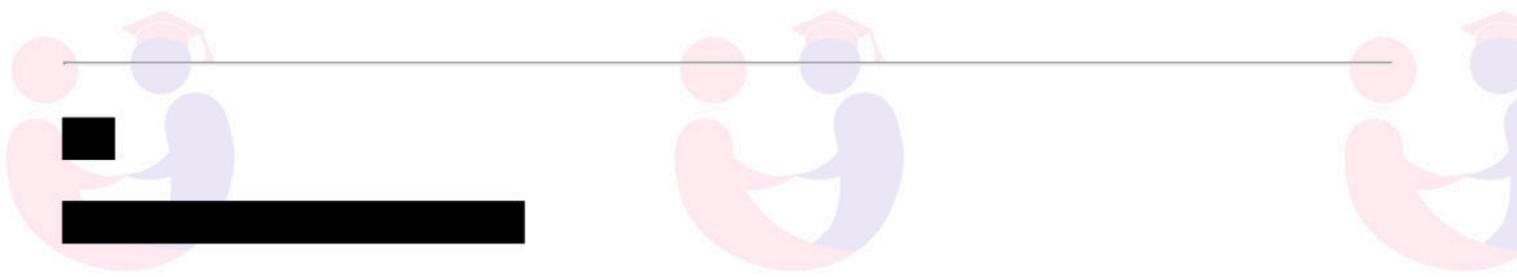
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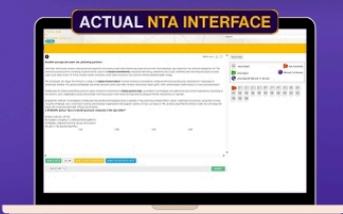
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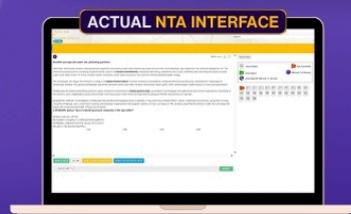
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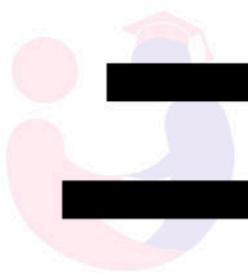
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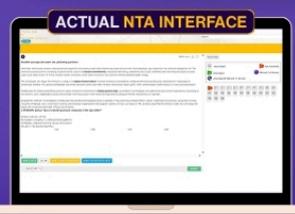


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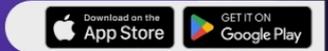
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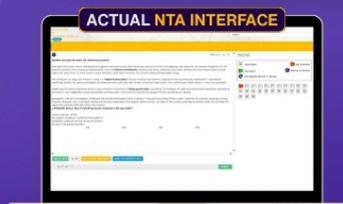
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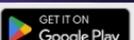
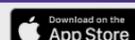
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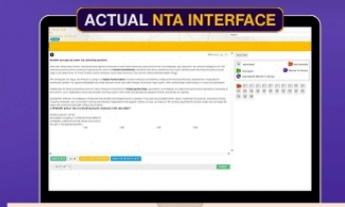
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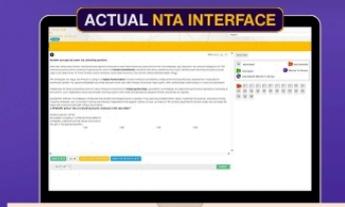
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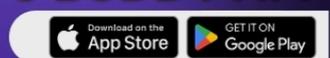
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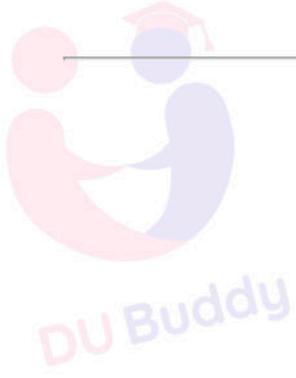


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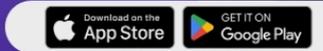
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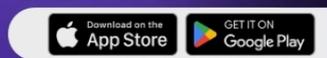
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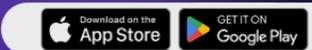
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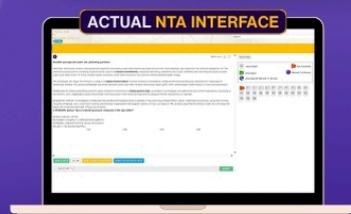
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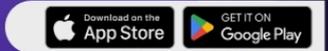
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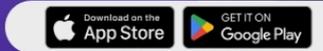
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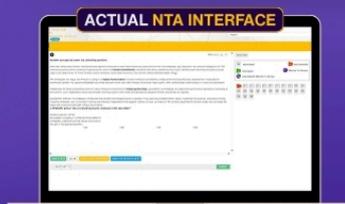
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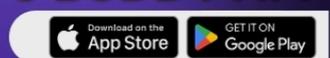
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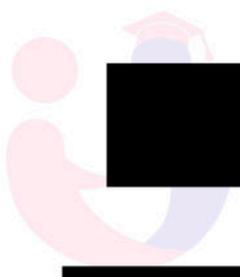
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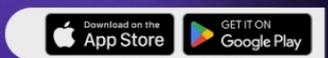
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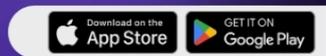
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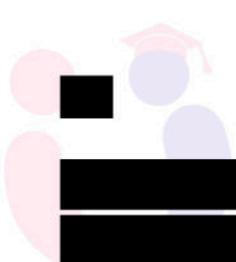
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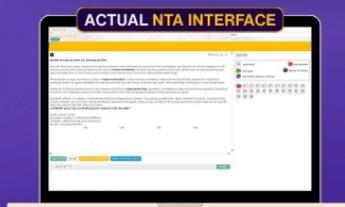
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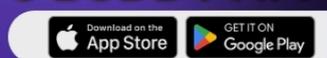
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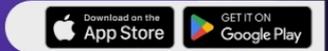
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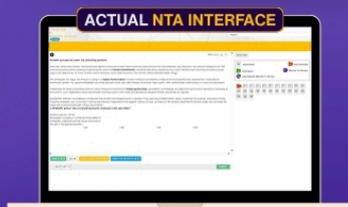
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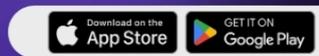
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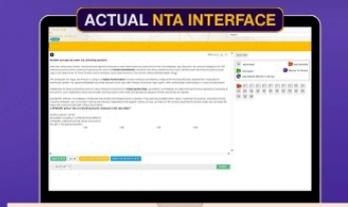
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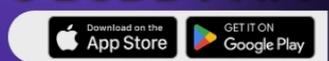
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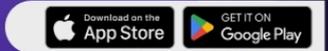
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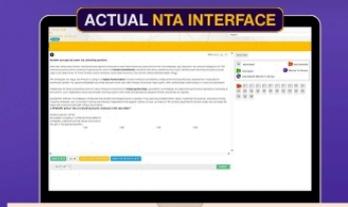
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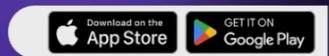
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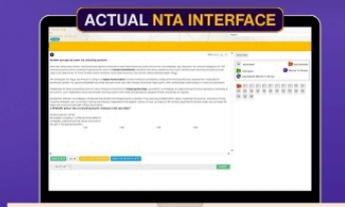
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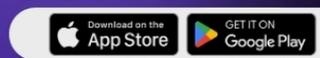
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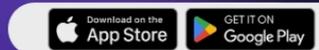
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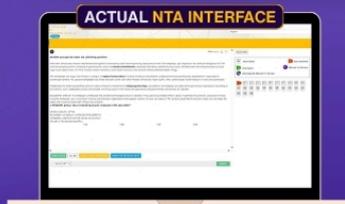
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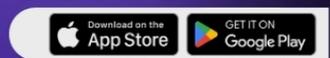
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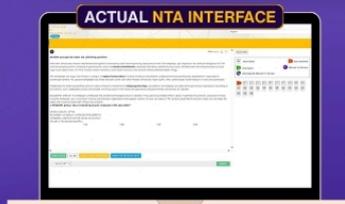
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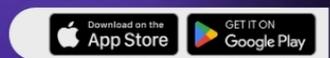
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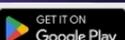
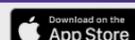
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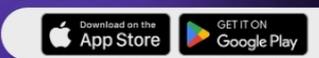
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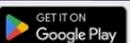
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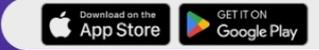
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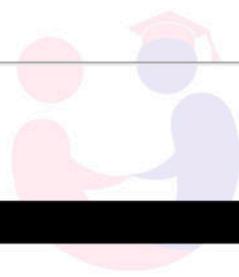
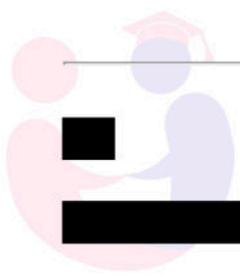
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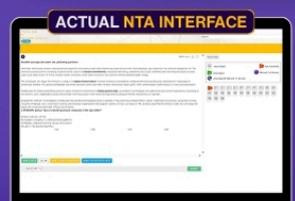


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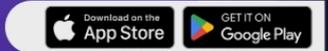
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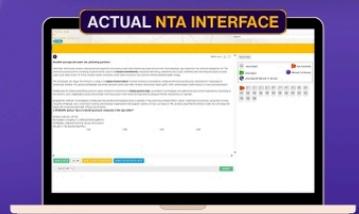
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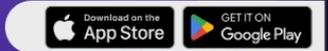
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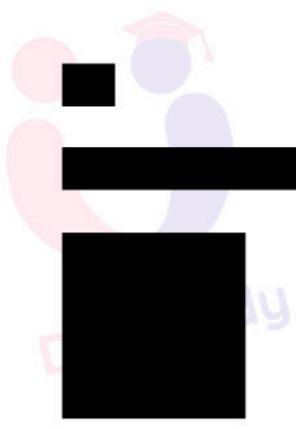
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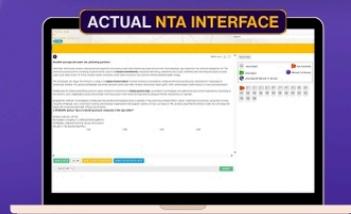


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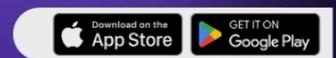
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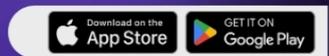
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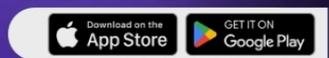
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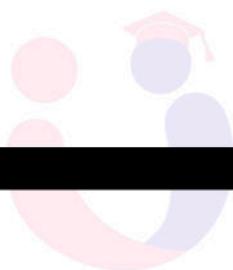
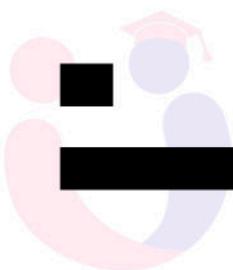
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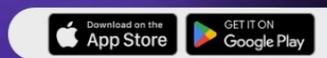
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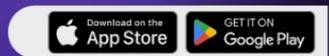
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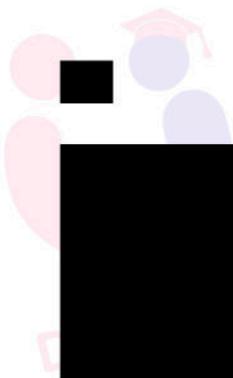
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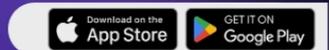
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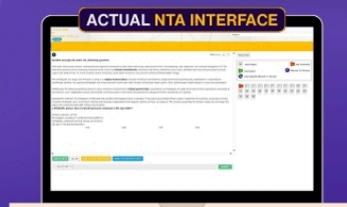
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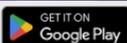
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